Fungal Products. Part XVIII.¹ ¹³C Nuclear Magnetic Resonance Spectrum and Biosynthesis of Colletodiol

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In the ¹³C n.m.r. spectrum of colletodiol (10,11-dihydroxy-2,8-dimethyl-1,7-dioxacyclotetradeca-4,12-diene-6,14dione) (1), the resonances have been assigned by correlation with ¹H n.m.r. assignments from irradiation experiments at single frequencies in the range of proton resonances. Spectra of colletodiol derived by incorporation of [1-¹³C]- and [2-¹³C]-acetate showed ¹³C enrichments consistent with a biosynthetic pathway from the union of a triketide and a tetraketide component.

COLLETODIOL, a metabolite of the plant pathogen, Colletotrichum capsici, possesses the structure (1).^{2,3} The following ¹³C n.m.r. studies have established the unexceptional biogenesis of this fungal product from acetate and indicate the pathway shown in the Scheme.

Since the chemical shifts of all the protons in colletodiol (1) have been previously assigned ² from ¹H n.m.r. spectra, the ¹³C resonances could be assigned by using the graphical method described by Birdsall *et al.*⁴ (Table 1). Plots of the ¹³C peak frequencies as a resonances of the 6- and 14-carbonyl carbon atoms remained constant in each irradiation experiment, providing an internal check.

For incorporation studies the fungus was cultured as described by MacMillan and Simpson³ except that mycelial growth was supported on a pad of glass wool. This arrangement facilitated the replacement of the original culture medium, after the production of colletodiol had ceased, by a medium containing only glucose and labelled acetate. In this way the previously



SCHEME Biosynthesis of colletodiol (1)

function of single frequencies in the range of proton resonances gave straight lines of similar slopes by leastsquares fit (FOCAL program). Care was taken to ensure that the irradiating frequencies were within the range where they were proportional to the residual coupling. Data from four experiments were used except where the irradiating frequency caused apparent complete decoupling of a particular ¹³C resonance. In such cases the data for the ¹³C peak at that frequency were too inaccurate to be included. The intersections of the straight line plots for each ¹³C peak gave values (Table 1) which were in good agreement with those previously determined² from the ¹H n.m.r. spectra. Clear-cut assignments of individual ¹³C resonances could therefore be made with the proviso that the assignments may be reversed for similar pairs of carbon atoms from each carbon chain. These assignments were entirely consistent with the literature values for ¹³C chemical shifts and with conventional proton decoupling data. The results from the graphical method were also self-consistent. For example the ¹³C chemical shifts derived from the intersections of the plots were in excellent agreement (Table 1) with those measured directly from the proton-decoupled ¹³C n.m.r. spectra. The ¹³C

reported ³ yield of colletodiol was increased and the related dilactones,³ colletol, colletal, and colletoketol, were formed in insignificant amounts.

Preliminary feeds with sodium $[2^{-14}C]$ acetate showed that glucose was required in the replacement medium; in its absence the yields of colletodiol and the percentage incorporation were very low. Following the approach of Holker *et al.*,⁵ the amount of ¹³C necessary for feeding was calculated by determining the overall dilution in feeds of $[2^{-14}C]$ acetate. When a replacement medium containing 5 μ Ci of sodium $[2^{-14}C]$ acetate, 500 mg of sodium acetate, and 50 g of glucose per litre was used, the overall dilution factor was 19, corresponding to a ¹³C enrichment of 4.7% for 90% ¹³C-labelled acetate.

The ¹³C enrichments at individual positions in $[1^{-13}C]$ and $[2^{-13}C]$ -acetate-derived samples of colletodiol (1) were calculated as described by Holker *et al.*⁵ However the results (Table 2) are expressed as enrichment factors (*i.e.* as the factor by which the natural ¹³C abundance has been increased) and not as ¹³C abundance in excess (*i.e.* the difference between enriched and natural ¹³C abundance) as used by Holker *et al.*⁵ The observed labelling pattern is in complete agreement with that required by the Scheme.

³ J. MacMillan and T. J. Simpson, J.C.S. Perkin I, 1973, 1487. ⁴ B. Birdsall, N. J. M. Birdsall, and J. Feeney, J.C.S. Chem. Comm., 1972, 316; B. Birdsall and J. Feeney, J.C.S. Perkin II, 1972, 1643.

¹ Part XVII, J. R. Bearder, V. M. Frydman, P. Gaskin, I. K. Hatton, W. E. Harvey, J. MacMillan, and B. O. Phinney, preceding paper.

² J. MacMillan and R. J. Pryce, Tetrahedron Letters, 1968, 5497.

⁵ J. S. E. Holker, R. D. Lapper, and T. J. Simpson, J.C.S. Perkin I, 1974, 2135.

Proton assignment 2-Me 8-Me	δ	н	δ	Carbon	
	Graphical method 1.50 ° 1.37 °	¹ H n.m.r. spectra 1.38 1.38	Graphical method 20.4 • 18.2 •	¹³ C n.m.r. spectra 20.4 18.2	assignment 2-Me 8-Me} c
$\frac{2}{8}$	5. 43 ° 5.22 °	5.43 • 5.30 5.22 • 5.16		68.7 68.3	$\binom{2}{8}c$
3 9	2.57 ° 1.83 °	$\begin{array}{c} 2.50,\ 2.21\\ 2.00,\ 1.5\end{array}$	41.2 ° 36.3 •	$\begin{array}{c} 41.2 \\ 36.2 \end{array}$	3 9
4 12	6.92 ^b 7.08 ^b	6.70 6.73	147.1 ^b 144.3	146.9 144.2	4 12} c
5 13	5.83 ^b 6.04 ^b	$\begin{array}{c} 6.72 \\ 6.14 \end{array}$	125.7 b 123.5 b	$\begin{array}{c} 125.7 \\ 123.5 \end{array}$	$\begin{bmatrix} 5\\13 \end{bmatrix} c$
10 11	3.84 ^b 4.12 ^b	$\begin{array}{c} 3.63 \\ 4.06 \end{array}$	71.8 ^b 73.8 ^b	71.8 73.8	10 11
				166.9 165.3	$\binom{6}{14}c$

TABLE 1 Assignment of ¹³C resonances of collection (1) from ¹H chemical shifts

^a Least-squares fit from 4 points. ^b Least-squares fit from 3 points. ^c Assignments of these pairs may be reversed without altering the biochemical conclusions.

TABLE	2
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¹³C N.m.r. data for colletodiol (1) derived from [1-¹³C]- and [2-¹³C]-acetates

Carbo n		Computer-listed intensities		Normalise to natural abundance		¹³ C Enrichment factor		
	δ _c	Natural	[1-13C] Ace- tate derived	[2-13C]Ace- tate derived	[1- ¹³ C]Ace- tate derived	[2-13C]Ace- tate derived	[1-13C]Ace- tate derived	[2- ¹³ C]Ace- tate derived
2	68.7	33.3	7.30	24.7	116.0	33.3	3.5	
8	68.3	29.5	7.74	24.3	123.0	32.8	3.8	
2-Me	20.4	24.2	1.74	56.2	27.6	75.7		2.7
8-Me	18.2	25.3	1.73	53.7	27.5	72.5		2.6
3	41.2	24.2	1.94	60.4	30.8	81.5		2.6
9	36.2	24.2	1.44	59.0	22.9	79.5		3.5
4	146.9	34.0	7.20	25.2	114.0	34.0	3.4	
12	144.2	37.0	7.90	26.2	125.0	35.4	3.5	
5	125.7	33.5	1.41	83.9	22.4	113.0		5.0
13	123.5	34.1	1.79	79.2	28.4	107.0		3.8
6	166.9	12.5	2.26	7.70	35.9	10.4	3.5	
14	165.3	9.8	3.62	8.42	57.5	11.4	5.0	
10	71.8	26.0	6.62	23.8	105.0	32.1	3.3	
11	73.8	29.3	1.62	72.6	25.7	98.0		3.8

EXPERIMENTAL

For general experimental procedures see ref. 3.

¹⁸C N.m.r. Determinations.—The ¹³C spectra were obtained for solutions in CDCl₃ with Me₄Si as internal standard by use of a JEOL PFT-100 instrument operating at 25.2 MHz in the Fourier Transform mode. Sweep widths of 6.25 KHz with 4 096 data points were used throughout to give ¹³C chemical shifts with an accuracy of ± 1.52 Hz (± 0.06 p.p.m.). A pulse width of 16 µs, 'tilt angle' of 45°, and repetition time of 0.73 s were used. Off-centre protondecoupled spectra were obtained by irradiation at 7.17 and 12.15 p.p.m. downfield from Me₄Si and at 2.84 and 7.8 p.p.m. upfield from Me₄Si at constant power.

Culture Methods.—The medium and conditions described by MacMillan and Simpson³ were used with the following modification. The vessels were packed with glass wool so that the mycelial pad was supported on the surface of the medium. ¹⁴C-Enriched Colletodiol.—Ten flat culture vessels each containing 200 ml of medium were inoculated with a spore suspension of *C. capsici*. After 12 days the medium was decanted and extracted with chloroform. The crude extract (235 mg) was purified by p.l.c. on silica gel developed with ethyl acetate-light petroleum (3:1). Colletodiol (115 mg) recovered from the band at $R_{\rm F}$ 0.3 was identical (m.p., n.m.r., g.l.c., and mass spectrum) with an authentic specimen.

In five of the above vessels the mycelium was re-incubated with replacement medium (200 ml per vessel) containing glucose (50 g l⁻¹), sodium [2-¹⁴C]acetate (5 μ Ci l⁻¹), and unlabelled sodium acetate (500 mg l⁻¹). After 10 days, colletodiol was isolated as before and crystallised to constant radioactivity (6.71 \times 10⁵ disint. min⁻¹ mmol⁻¹; yield 45 mg).

Sodium $[1^{-13}C]$ Acetate Feed.—By the same procedure as for sodium $[2^{-14}C]$ acetate, the first incubation gave

unlabelled colletodiol (60 mg). Incubation of the mycelium from five vessels with replacement medium containing glucose (50 g l⁻¹) and $[90\%-1^{-13}C]$ acetate (500 mg l⁻¹) gave $[1^{-13}C]$ acetate-derived colletodiol (18 mg).

[1-13C]acetate-derived colletodiol (18 mg). Sodium [2-13C]Acetate Feed.—From an identical experiment with [90%-2-13C]acetate, unlabelled colletodiol (112 mg) and $[2\mathchar`-1\mathc$

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